

SAULT COLLEGE OF APPLIED ARTS & TECHNOLOGY  
SAULT STE. MARIE, ONTARIO

COURSE OUTLINE

COURSE TITLE: AIR MONITORING ANALYSIS

CODE NO.: ENV 303 - 5 SEMESTER: VI

PROGRAM: ENVIRONMENTAL ENGINEERING TECHNOLOGY

AUTHOR: L. VANDERZWET

DATE: APRIL 1994 PREVIOUS OUTLINE DATED: MAY 1991

APPROVED: \_\_\_\_\_  
DEAN DATE



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COURSE NAME

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COURSE NUMBER

TOTAL CREDIT HOURS: 5

PREREQUISITE(S): CHM221

**I. PHILOSOPHY/GOALS:**

Air Monitoring Analysis expands on topics covered in the first instrumental course (CHM221). As well, other relevant electrochemical methods for analysis will be studied. Other major goals include developing the ability to determine the appropriate procedures for the collecting, storage and sampling of air, soil and water samples.

**II. STUDENT PERFORMANCE OBJECTIVES:**

Upon successful completion of this course the student will be able to:

1. Describe the theory of electrochemistry and electrochemical cells, including half cells.
2. Understand the different types of reference and indicator electrodes and electrometric methods of analysis, including pH meters, specific ion electrodes and polarography.
3. Determine the proper sampling and storage procedures for air, soil and water samples, based on sample size, required determinations, homogeneity of sample, possible sources of contaminants, etc.
4. Determine the best method of sample preparation and analysis for air, soil and water samples, based on sample size, required determination, concentrations, costs, etc.

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III. TOPICS TO BE COVERED:

a) THEORY

Hours	Topics
1	1. Course Introduction
4	2. Review of Electrochemical Theory
6	3. Electrometric Methods of Analysis
8	4. Sampling, Storage and Analytical Procedures of Atmospheric pollutants
7	5. Sampling and Storage Procedures of Soil and Water Samples
6	6. Choosing the best method of analysis
5	7. Tests/Exam

b) LABORATORY

The student will complete three water analysis experiments designated for this course, in the allotted time. A major component of these experiments will be sample preparation. The exact experiments will depend on those completed in CHM221 but can include:

- Specific Ion Electrode ( $F^-$ ,  $NO_3^-$ )
- Gas Chromatography (solvents)
- Atomic Absorption (heavy metals)
- Spectronic Colourimeter ( $Co^{+3}/Cr^{+3}$ )
- HPLC (anions)
- Titroprocessor (acidity/alkalinity)
- Kjeldahl (nitrogen)
- Double beam spec (phosphate/phenol)
- Total Carbon

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III. TOPICS TO BE COVERED: (continued)

b) LABORATORY

The student will be able to subject their results to statistical analysis and determine:

1. Precision
2. Relative Error
3. Average Deviation
4. Standard Deviation
5. Whether a result should be excluded by the 2.5d rule, 4.0d rule and the Q test.

The student will also complete a group project where atmospheric pollutants will be collected with the "Hi-Vol" and analysed for total particulates, organics/inorganics particulates and specific metals. As well, the report will include methodology, results and discussion of an extraction for oils from a soil sample and of collecting various gases and vapours within the laboratory environment.

IV. EVALUATION METHODS:

Grades:	> 90%	- A+
	80 - 89%	- A
	70 - 79%	- B
	60 - 69 %	- C
	< 60%	- R

The final grade is arrived at by totalling the theory marks (50%) and the lab marks (50%).

The theory mark is the sum of all tests, assignments and final examinations. Assignments and labs are due on the date specified and late assignments and labs will not be accepted. The lab mark is the sum of all marks awarded for the analysis plus the written report for each of the three experiments. The group report will have the weighting of two regular labs. The analysis is graded on accuracy and precision while the lab report is graded on format, content and neatness.

Your grade will be greatly affected by attendance at scheduled classes and labs. Eighty-five percent attendance is required at all theory classes while 100% is needed for all labs.

V. REQUIRED STUDENT RESOURCES:

**Textbook:**

Braun, Robert, D., *Introduction to Chemical Analysis*, McGraw Hill, 1982.

**Supplemental Texts and References:**

A number of texts on analytical chemistry are available in the school library.

VI. SPECIAL NOTES:

Students with special needs (e.g. physical limitations, visual impairments, hearing impairments, learning disabilities) are encouraged to discuss required accommodations confidentially with the instructor.

Your instructor reserves the right to modify the course as he/she deems necessary to meet the needs of students.